PII: S0040-4039(96)02433-1

Desymmetrisation of Bicyclic, Meso-Anhydrides by Proline Esters

Iwan G. Jonesa, Wyn Jonesb, Michael Northax, Marta Teijeirac, and Eugenio Uriartec

a) Department of Chemistry. University of Wales, Bangor, Gwynedd, U.K., LL57 2UW; Email m.north@bangor.ac.uk

b) Peboc Division of Eastman Chemical (UK) Ltd., Industrial Estate, Llangefni, Anglesey. Gwynedd, LL77 7YQ
c) Department of Organic Chemistry, University of Santiago de Compostela, Spain

Abstract: The desymmetrisation of cyclopropane and cyclopentane derived, bicyclic meso-anhydrides using proline esters provides a versatile approach to the synthesis of enantiomerically pure amido acids which can subsequently be converted into β -amino acid containing peptide analogues. © 1997, Elsevier Science Ltd. All rights reserved.

In recent publications, we have reported the facile desymmetrisation of tricyclic, meso-anhydrides derived from norbornane using methyl (S)-prolinate as shown in Scheme 1. This chemistry has allowed us to access enantiomerically pure amido acids which are versatile building blocks for the synthesis of conformationally constrained peptide analogues² and also have other potential applications. The asymmetric induction observed during this process seems to depend upon the steric difference between the two faces of the tricyclic anhydride. Attempts to extend this chemistry to bicyclic anhydrides derived from cyclohexane-1,2-dicarboxylic acid resulted in no asymmetric induction, and cyclopropane dicarboxylic anhydride gave only an inseparable mixture of diastereomers in 3:1 ratio. In the present paper, we report the successful desymmetrisation of bicyclic anhydrides derived from cyclopropane and cyclopentane anhydrides using proline esters.

Scheme 1

In an attempt to improve the diastereoselectivity of the desymmetrisation of cyclopropane anhydride 1, the use of *tert*-butyl (S)-prolinate 2 was investigated (Scheme 2). We reasoned that the larger ester substituent on the proline would be better able to discriminate between the two faces of anhydride 1, and so give a better

asymmetric induction. Surprisingly however, the desired amido acid **3a,b** was obtained in 57% yield as a 3:1 ratio of diastereomers, the same ratio that had been obtained using methyl (S)-prolinate. In this case however, the two diastereomers of acid **3** were readily separable, as trituration with diethyl ether removed the minor isomer, leaving acid **3a** as a white crystalline solid in 40% isolated yield. The absolute configuration of the major diastereomer **3a** was determined to be as shown in **Figure 1** by X-ray crystallography.

Amido acid 3a proved to be a useful starting material for the synthesis of pseudo-peptides incorporating a cis-2-amino-cyclopropane carboxylate unit. Thus Curtius rearrangement⁵ of acid 3a gave isocyanate 4 in 67% yield. Reaction of isocyanate 4 with tert-butyl (S)-prolinate 2 or HN-(S)-Pro-(S)-Phe-(S)-Phe-OMe gave β -amino cyclopropane carboxylate containing pseudo-peptides 5 and 6 in 70 and 74% yields respectively as shown in Scheme 2. However, all attempts to convert isocyanate 4 into the corresponding amine were unsuccessful, resulting only in decomposition. Reaction of the isocyanate with β -(trimethylsilyl)ethanol did produce urethane 7 in quantitive yield, however attempted removal of the β -(trimethylsilyl)ethoxycarbonyl protecting group with tetrabutylammonium fluoride resulted in concomitant ring opening of the cyclopropane to give aldehyde 8 in 65% yield.

Figure 1: ORTEP Diagram of Acid 3a

The desymmetrisation of cis-1,2- and cis-1,3-cyclopentane dicarboxylic anhydrides⁷ (9 and 10 respectively was also investigated as shown in Scheme 3. Anhydride 9 gave a 3:1 ratio of diastereomers 11a, 12a when desymmetrised using methyl (S)-prolinate, and a 4:1 ratio of diastereomers 11b, 12b when treated with tert-butyl (S)-prolinate. These results are very similar to those obtained with cyclopropane anhydride 1. The major isomer in these reactions was identified as 11a,b, by conversion of the amido acids into lactone 13 by esterification using dimethyl sulphate, followed by chemoselective reduction with lithium borohydride and

acidic cyclisation of the resulting amido alcohols. Lactone 13 was found to possess a negative specific rotation and so to be the enantiomer of a known lactone, 8 and hence to have the stereochemistry shown in structure 13.

In the case of the 1,3-anhydride 10, a single diastereomer of amido acid 14a,b was obtained using either methyl or tert-butyl (S)-prolinate. The much better asymmetric induction obtained with anhydride 10 is probably due to the similarity between the structure of this anhydride and that of the tricyclic norbornane derivatives investigated previously. Unfortunately, all attempts to convert amido acids 14a,b into the corresponding lactone were unsuccessful. However, the stereochemistry of the amido acid is assumed to be as shown in structure 14a,b, based upon analogy with the desymmetrisation of structurally related norbornene anhydrides.

In conclusion, the desymmetrisation of bicyclic, *meso*-anhydrides using chiral, proline esters can proceed with useful to excellent asymmetric induction, and the adducts can be transformed into β -amino acid containing peptides. Our work on the scope and applications of this chemistry is continuing, and further results will be published in due course.

Acknowledgements

The authors thank the EPSRC and Peboc Division of Eastman Chemical (UK) Ltd. for a CASE award studentship to IGJ. X-ray structures were determined by the staff of the EPSRC X-ray crystallography service at the University of Wales, Cardiff, and mass spectra were recorded by the EPSRC mass spectrometry service at the University of Wales, Swansea.

References and Notes

- 1. North, M.; Zagotto, G.; Synlett, 1995, 639; Albers, T.; Biagini, S.C.G.; Hibbs, D.E.; Hursthouse, M.B.; Malik, K.M.A.; North, M.; Uriarte, E.; Zagotto, G.; Synthesis, 1996, 393.
- Jones, I.G.; Jones, W.; North, M.; Synlett, accepted for publication.
- 3. All new compounds gave satisfactory spectroscopic data (¹H NMR, ¹³C NMR, IR, m/z, and HRMS).
- Full details of the X-ray analysis will be published elsewhere.
- Canone, P.; Akssira, M.; Dahdouh, A.; Kasmi, H.; Boumzebra, M.; Tetrahedron, 1993, 49, 1985; Furuta, K.; Hayashi, S.; Miwa, Y.; Yamamoto, H.; Tetrahedron Lett., 1987, 28, 5841; Arai, Y.; Kawanami, S.; Koizumi, T.; J. Chem. Soc., Perkin Trans. 1, 1991, 2969; Zegaff, C.; Poncet, J.; Jouin, P.; Dufour, M-N.; Castro, B.; Tetrahedron, 1989, 45, 5039.
- Similar ring opening reaction of cyclopropylamines bearing an electron withdrawing substituent on the cyclopropane ring have previously been reported. Cannon, J.G.; Garst, J.E.; J. Org. Chem., 1975, 40, 182; Augustine, R.L.; Pinto, F.G.; J. Org. Chem., 1975, 40, 115.
- 7. Hronowski, L.J.J.; Szarek, W.A.; Can J. Chem., 1988, 66, 61.
- 8. Bridges, A.J.; Sundara, R.P.; Ng, G.S.Y.; Jones, J.B.; J. Am. Chem. Soc., 1984, 106, 1461.

(Received in UK 25 November 1996; revised 10 December 1996; accepted 13 December 1996)